

cis-Vinylphosphonates and 1,3-Butadienylphosphonates by Zirconation of 1-Alkynylphosphonates

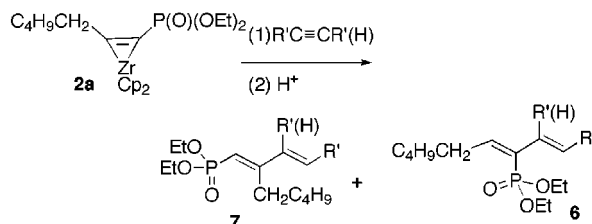
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ABSTRACT



Addition of "zirconocene" to 1-alkynylphosphonates gives three-membered zirconacycles that can be hydrolyzed to *cis*-vinylphosphonates or further reacted with alkynes/hydrolysis to give substituted (*Z,E*)- and (*E,E*)-1,3-butadienylphosphonates.

Although 1-alkynylphosphonates have been known since 1957 and their synthesis was developed in the 1960s,¹ addition reactions of organometallics remain relatively unexplored and include *syn* addition of organocuprates to 1-alkynylphosphonates to give 2,2-disubstituted vinylphosphonates,² reaction of α -stannylated phosphonates with aldehydes to give *E/Z* mixtures of 1,2-disubstituted vinylphosphonates,³ *anti* hydrotelluration of 1-alkynylphosphonates,⁴ Heck reactions using aryldiazonium salts,⁵ α -lithiation of β -oxy or β -thio vinylphosphonates,⁶ NaH-catalyzed olefination of benzenesulfinylmethylphosphonates,⁷ and addition of sodium organyl chalcogenolates to 1-alkynylphos-

phonates.⁸ These reactions provide access to 1-alkenylphosphonates that are very useful compounds for organic transformations⁹ and for the synthesis of biologically active compounds.¹⁰ We have recently started to investigate the addition of organometallic reagents to 1-alkynylphosphonates. Thus, we have discovered that the hydroboration of 1-alkynylphosphonates can be controlled to place boron on either C1 or C2 of the triple bond by proper use of base,

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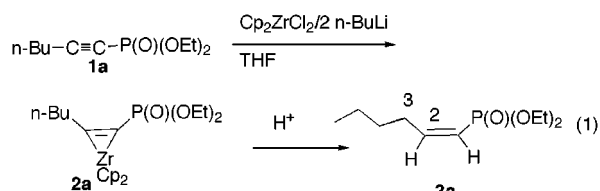
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catalyst, and reaction time. Another very useful reaction of triple bonds that has not been applied to 1-alkynylphosphonates is the addition of zirconocene. Hydrolysis of the zirconacycles would provide *cis*-vinylphosphonates, and subsequent insertion reactions of the three-membered zirconacycles with various electrophiles would provide access to functionalized vinylphosphonates.¹¹ In the context of this paper, we explored the insertion of alkynes to provide substituted 1,3-butadienylphosphonates.

cis-Vinylphosphonates are useful intermediates in organic transformations.^{9a} Reduction of 1-alkynylphosphonates with hydrogen under various conditions has been reported. Generally, mixtures of *cis*- and *trans*-vinylphosphonates are obtained.¹² When diethyl butynylphosphonate¹³ was treated with Cp₂ZrCl₂/2*n*-BuLi,¹¹ and hydrolyzed, GCMS analysis indicated complete conversion. *cis*-Diethyl 1-butenylphosphonate was obtained as a single isomer in 76% isolated yield (eq 1).¹⁴



The reaction is general, isolated yields are excellent, and only the *cis*-vinylphosphonates, **3**, are obtained (Table 1).

Table 1. Synthesis of **3** and Selected NMR Data

entry	R	yield, ^a %	<i>J</i> (Hz)	
			<i>J</i> _{P-H2}	<i>J</i> _{P-C3}
a	C ₄ H ₉	79	53.1	8
b	C ₅ H ₁₁	76	53.1	8
c	ClC ₃ H ₆	63	52.5	8.3
d	Ph	76	51.5	8.8
e	TBDMSOC ₃ H ₆	78	52.6	8.1

^a Isolated yields. GCMS conversion was >99%, except for entry **e**, 85%.

The stereochemistry of the **3** was determined from the coupling constants obtained from the NMR data where both the ³*J*_{P-H2} coupling constant (~50 Hz) and the small ³*J*_{P-C3} (~8 Hz) indicates that H2 is *trans* to phosphonate group and that the R group is *cis*. Table 1 shows select coupling

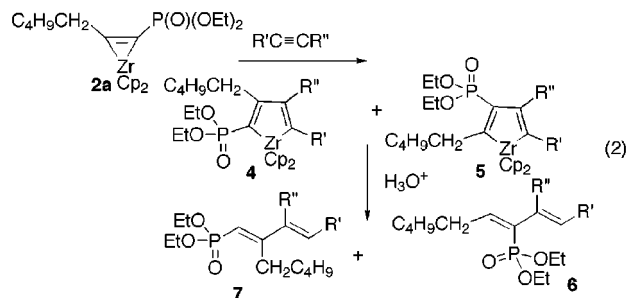
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constants of vinylphosphonates obtained by reaction of 1-alkynylphosphonates with “zirconocene” followed by hydrolysis.

A further utilization of the three-membered zirconacycles, **2**, in this work was the investigation of alkyne insertion. This would lead to 1,3-butadienylphosphonates. The latter are interesting compounds that undergo a variety of reactions including 1,3-additions,¹⁵ cycloaddition with CH₂N₂,¹⁶ and [2 + 2] cycloadditions.¹⁷ They have been prepared by isomerization of 1-alkynylphosphonates in the presence of palladium salts,¹⁸ by Knoevenagel reaction,¹⁷ by reaction of unsaturated cyanophosphonates with *N*-tosylsulfonylimines,¹⁹ and by procedures similar to the preparation of vinylphosphonates.^{9a}

When zirconacycles **2** (R = C₄H₉CH₂) were treated with a different alkyne (both terminal and internal alkynes were used), and the reaction mixture was hydrolyzed, two isomeric products were detected by GCMS, **6** and **7**, and isolated by silica gel chromatography (eq 2). Presumably, they arise from



zirconacycles **5** and **4**. Other possible isomeric 1,3-butadienylphosphonates were not isolated, apparently due to unfavorable steric interactions between the R' groups of the incoming alkyne, the phosphonate, and C₄H₉CH₂ groups of the zirconacycle.

Results are listed in Tables 2 and 3. With terminal alkynes, **6** was the major isomer in all cases, apparently due to steric considerations. With an internal alkyne (Table 2, entry e),

Table 2. Synthesis of **6** and Selected NMR Data

entry	R''	R'	yield ^a %	<i>J</i> (Hz)		
				<i>J</i> _{H4-H5}	<i>J</i> _{P-H2}	<i>J</i> _{P-C3}
a	H	C ₄ H ₉	68	15.4	48.8	5.4
b	H	C ₅ H ₁₁	60	15.7	48.9	5.6
c	H	Ph	73	16.2	48.9	6.5
d	H	C ₃ H ₆ Cl	57	15.9	54.0	6.6
e	C ₂ H ₅	C ₂ H ₅	<3% ^b			

^a Isolated yield GCMS conversion for combined **6** and **7** was >99%.
^b Not isolated.

compound **7e** was essentially the only product isolated (Table 3), **6e** being obtained in less than 3%. The structures of compounds **6** were determined by NMR spectroscopy. The doublet of triplets in the double bond region 6.5–6.2 ppm indicates that the alkyne coupling occurred on C1. Also, the

Table 3. Synthesis of **7** and Selected NMR Data

entry	R''	R'	yield, ^a %	J (Hz)		
				J _{H4-H5}	J _{P-C3}	J _{P-C4}
a	H	C ₄ H ₉	15	15.9	5.6	28
b	H	C ₅ H ₁₁	19	15.8	5.6	27.8
c	H	Ph	11	17.2	5.7	26.7
d	H	C ₃ H ₆ Cl	20	15.9	6.4	26.6
e	C ₂ H ₅	C ₂ H ₅	83		6.5	23

^a Isolated yields. GCMS conversion for combined **6** and **7** was >99%.

large ³J_{P-H2} coupling constant (~50 Hz) and the relatively small ³J_{P-C3} coupling constant (5–6 Hz) indicate that the stereochemistry of the C1–C2 double bond is *Z*. The

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(14) To 1 mmol (0.292 g) of zirconocene dichloride dissolved in 5 mL of dry THF was added 2 mmol (1.25 mL of *n*-BuLi 1.6 M in hexane) dropwise at –78 °C. The mixture was stirred for 3 h, then 0.9 mmol of 1-alkynylphosphonate was added, and the mixture was slowly warmed to 25 °C and stirred overnight. The mixture was worked up with dilute aqueous HCl, and the vinylphosphonate was extracted in ether and separated on a silica gel column (80% petroleum ether: 20% ethyl acetate).

(15) Martin, S. F.; Garrison, P. J. *Synthesis* **1982**, 394.

coupling constants ³J_{H4-H5} (15–16 Hz) indicate that the hydrogens are *trans*. Thus compounds **6** have a *Z,E* configuration for the double bonds.

In a similar manner the structures of compounds **7** were determined. The doublet in the region 5.3–5.5 ppm corresponds to the hydrogen on the C1. This indicates that the alkyne coupling occurred on C2. Also, the large ³J_{P-C4} coupling constants (23–28 Hz) and the relatively small ³J_{P-C3} coupling constant (5.6–6.5 Hz) indicate that the stereochemistry of the C1–C2 double bond is *E* in compounds **7**. In addition, the multiplet (H4–H5) in the region 6.3–5.9 ppm corresponds to two vinylic hydrogens with a coupling constant ³J_{HH} (15.8–17.2 Hz) indicative of the *E* configuration. Thus compounds **7** are the *E,E* isomers (Table 3).

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Supporting Information Available: Experimental procedures and full NMR data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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